

# Experience of Degradation of Mooring Systems used in the



# Mooring Inspections - Worldwide

**20 Years** experience in the United Kingdom, Norway, Canada, Gulf of Mexico, West Africa, Australia, South East Asia, and India.

● Welaptega Global Projects

## Agents:



IEV Group



Accpron



Beijing Safetech



GCA Energy



# Our Valued Clients

Welaptega operates globally with clients and projects in just about every region of the offshore oil and gas industry.



# Problems:

- Mooring components:
  - Equipment failing prematurely
  - Galvanized shackles corroded/worn within weeks vs months/years in other environments
- Power Cables:
  - Armour wires worn away and eventually break
  - Characterized by armour wires forming sharp points; always shiny/bright

# Mooring Component Failure

- Galvanized shackles corroded/worn prematurely.
- Obviously a combination corrosion and wear. But which is dominant?



American Bureau of Shipping



Det Norske Veritas

# Wear

## Wear between components expressed by Archard Equation:

- Wear (Volume) =  $K \times \text{Work Done}$
- Work Done = Frictional Force  $\times$  Distance Travelled

- K is a function of:

**Contact Pressure**

**Lubrication**

**Surface Roughness**

**Sliding Velocity**

**Surface Hardness**

**Ambient Temperature**

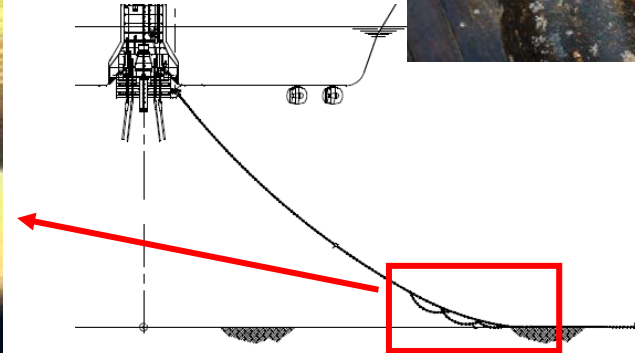
**Rate of Dissipation of Frictional Heat**

**Elastic and Shear Moduli**



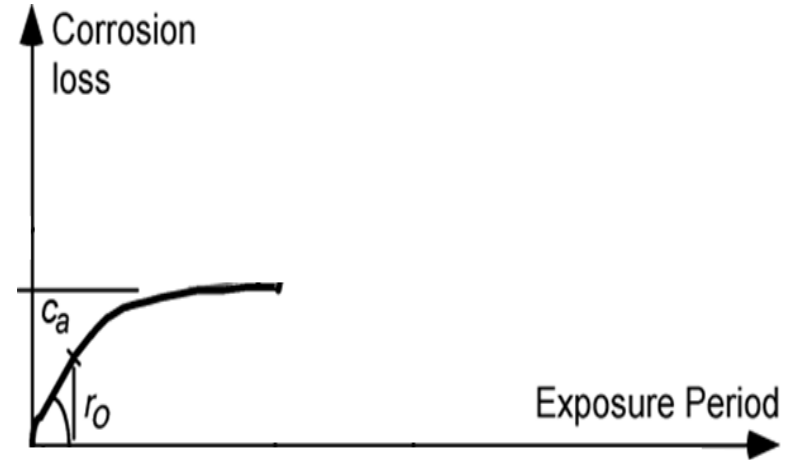
# Wear

- Wear rate highest where there high tension and relative motion; relative motion dominates
- Reduce relative motion → reduce the wear
- However, it's unlikely that any of these factors are unique to the Minas Passage



# Corrosion

- High initial rate of corrosion ( $r_0$ ), slowing to a lower steady-state rate as oxide layers accumulate to reduce oxygen diffusion to steel surface
- Major contributors to corrosion rate:
  - Dissolved oxygen content
  - Water temperature
  - Flow rate, water velocity
    - Significant impact on corrosion rate in early stage of exposure to sea water ( $r_0$ )
    - Once oxide layers formed, flow rate is insignificant on steady state corrosion rate



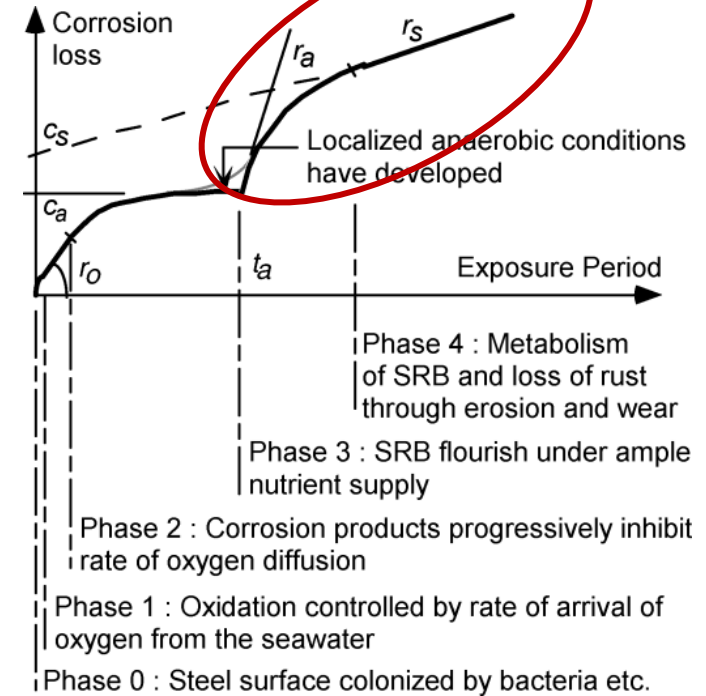
Robert E. Melchers

Development of new applied models for steel corrosion in marine applications including shipping



# Multiphase Aerobic-Anaerobic (Biological) Corrosion

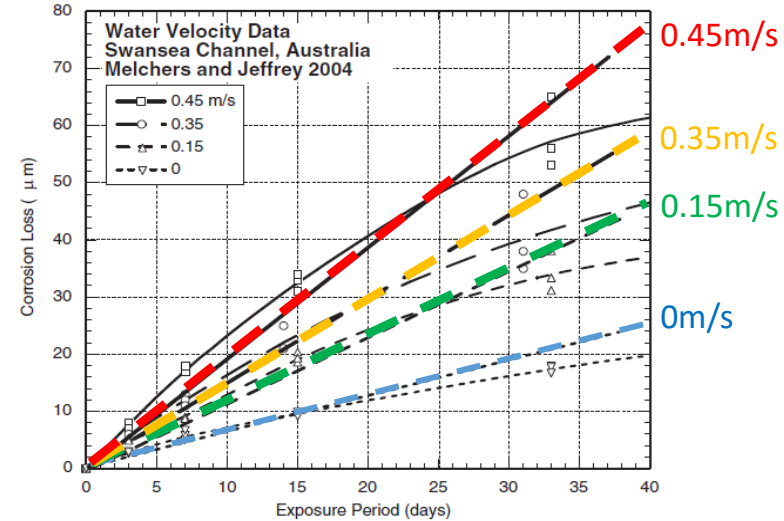
- Microbes (especially Sulfate Reducing Bacteria) do not attack the steel – metabolic by-products create a corrosive environment
- Oxide layers and biofilm must eliminate oxygen diffusion (anaerobic condition)
- Very aggressive corrosion rates in warm water especially under ample supply of nutrients – nitrogen



Robert E. Melchers  
Development of new applied models for steel corrosion in marine applications including shipping

# What's unique about the Bay Of Fundy wrt Corrosion Rates?

- Water Temperature?
  - Pretty typical of other environments without low corrosion rates
- Dissolved Oxygen?
  - Any evidence of this being higher than typical levels at similar water depths?
- Microbiological Corrosion?
  - High nutrient levels from agricultural run-off could be a factor
  - More typical of warm water environments; Aerobic corrosion layer must build up to create anaerobic environment
- Water Flow Rate
  - Phase 1 corrosion rate increases dramatically with flow rate
  - Note: test shown only goes up to 0.45m/s
    - Peak Tidal current at Cape Sharp is 5.5m/s

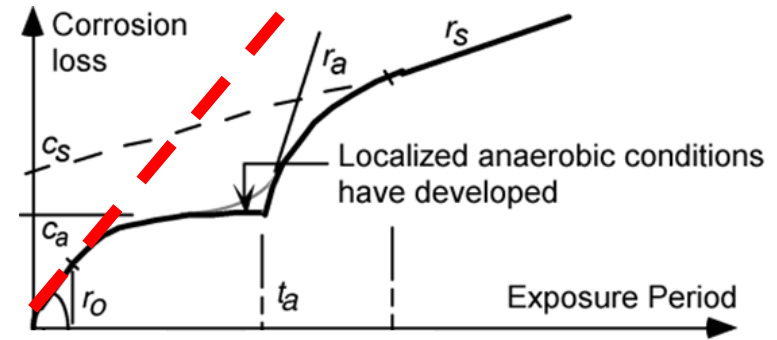


Average water temperature of 20-deg C; higher than Bay of Fundy, but relationship between flow rate and corrosion rate remains the same.

Robert E. Melchers  
Corrosion of working chains  
continuously immersed in seawater

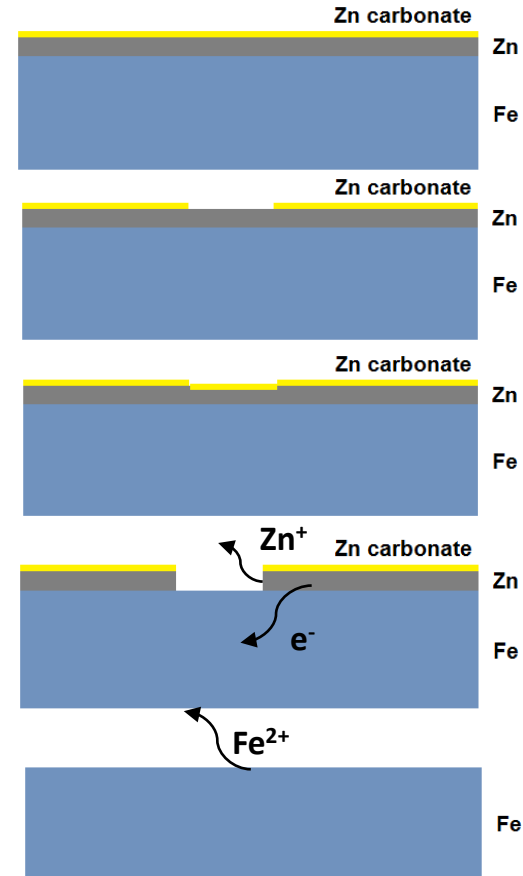
## *Possible Cause*

- High flow rate is leads to high initial corrosion rate, and
- Oxide layer needed to slow the corrosion rate does not establish due to abrasion:
  - Between mooring components,
  - Between components and seabed,
  - Abrasion due to entrained sand in flow?
- Components continually remaining in the Phase 1 high aerobic corrosion phase



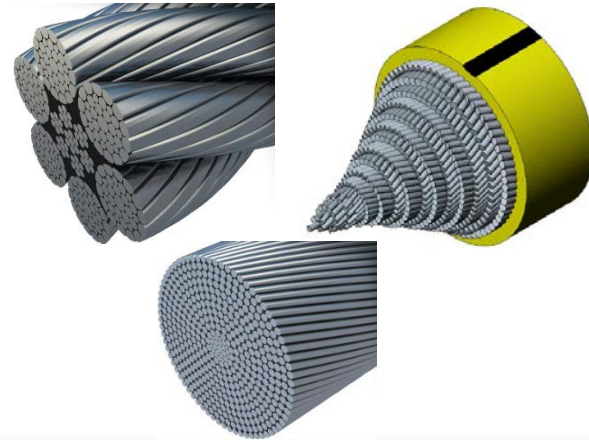
# What about Galvanizing?

- In sea water, galvanized surfaces form zinc carbonate and calcareous deposits to form a barrier to corrosion
- The zinc carbonate barrier is self repairing; if damaged, new zinc will be consumed to repair the barrier
- However, if the barrier is continually damaged, zinc will be consumed quickly
- If all zinc steel surface is exposed, zinc is further consumed due to creating of a galvanic corrosion cell
- Contributing factors lowering performance of galvanizing:
  - Mechanical damage to barrier – wear between components
  - Water temperature
  - **Flow rate/abrasion**
    - Removes passive scales and causes further zinc consumption



# Galvanizing

- In oil and gas industry:
  - MOORING CHAINS and SHACKLES are **never** galvanized
    - Steel grades used for O&G platforms are higher strength than typical marine grade; require heat treatment
    - Galvanizing process requires high heat; would reduce material properties
    - Chains designed with corrosion allowances instead
  - WIRE ROPE MOORINGS are **always** galvanized
    - 6-strand wire or
    - Torque-balanced spiral strand (sheathed or unsheathed)
    - Not typically used in areas of high wear for long periods of time



# Power Cable Failure

- Outer armour wires formed into **sharp points** and lead to failure
  - Cables are not buried (sediment has been washed away in high flow)
  - Cable laid on hard rock seabed, but failure mode is unrelated to the cable wearing from movement on the seabed itself
  - Armour wires are always shiny/bright
- Cables exhibiting this deterioration were move to areas of lesser current



# Wire Rope Corrosion

- Drilling rig wire rope after 7 years in the North Sea
- Aside from corrosion, strands appear in otherwise good condition
  - Zinc is fully consumed; steel stands freely corroding
- Wire wet sand-blasted to inspect further



Degradation of wire rope mooring lines in SE Asian waters; Chaplin, Potts, Curtis

# Wire Rope Corrosion

- Once the zinc is consumed, the bare wires freely corrode where zinc carbonate layer is damaged
- Corrosion is accelerated on the side of the wires in close proximity to adjacent strands
- Small movements between strands will disturb and wear away at the corrosion layer, causing further corrosion to the side of strands
- Crown of wires are not abraded and still protected by zinc-carbonate deposits
- The strands eventually appear rectangular or fin-like, forming sharp points



# Power Cable – Moving to area of lesser current solved problem

- Lower current area will improve performance of galvanizing layer and decrease the steel corrosion rate if zinc is eventually fully consumed
- Note - Armour wires in failed power cables said to have looked bright/shiny
  - Photo of wire rope (right) has been wet sand-blasted
  - Possible the entrained sand in high flow environment removes corrosion products, contributing to high corrosion rate



# *Possible* Solutions

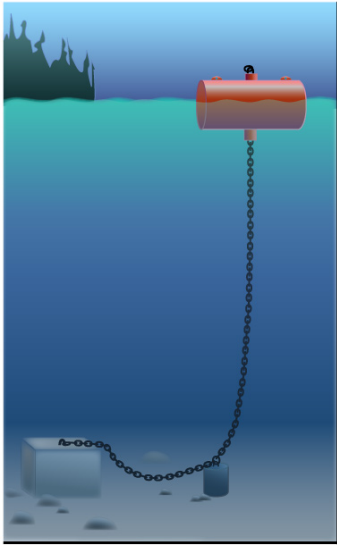
Assuming moorings and power cables cannot be moved to lower-flow areas, consider:

- Reducing motion between mooring components
- Better coatings or design with high corrosion/wear allowance
- Corrosion resistant or non-corroding materials

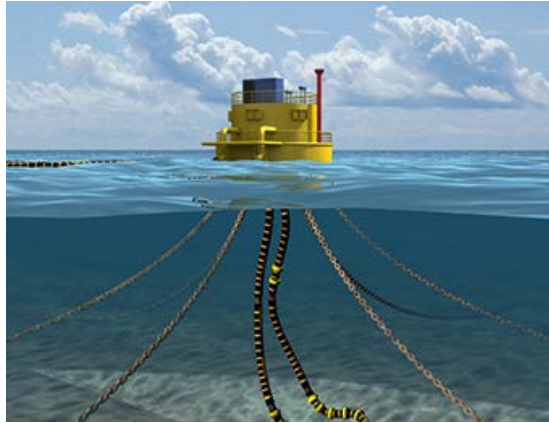


# *Possible* Solutions to mooring failure

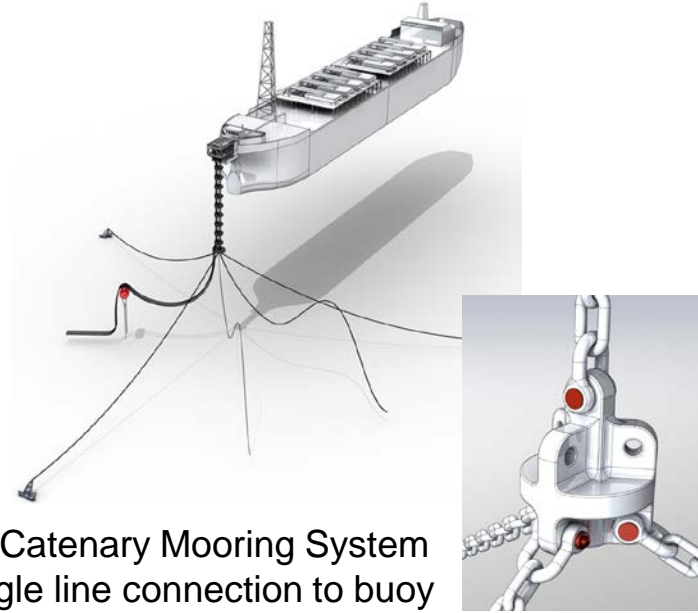
- Reduce motion between mooring components to reduce wear



Single line  
Low pretension  
High relative motion



Typical Single Point Mooring for  
O&G industry  
6 Lines; each line pretensioned to  
limit lines slackening and relative  
motion



Tri-Catenary Mooring System  
Single line connection to buoy  
Split to 3 lines in mid-water  
Maintains minimum tension on all lines



# *Possible* Solutions to mooring failure

Coatings or high corrosion/wear allowances

- Coatings for Chain, Steel Connectors
  - TSA – Thermal Sprayed Aluminum
    - Similar to galvanizing, but provides better protection in marine environments
    - Aluminum provides anodic protection if surface coating is damaged
    - Low heat process – suitable for high strength steels
  - TSC – Thermal Sprayed Carbide
    - Resistant to wear, even at the contact zone between chain links
  - More details needed of how these coatings perform, especially in high-flow environments
- Add a corrosion/wear allowance to the required chain size
  - Use recovered components to estimate corrosion/wear rate
  - Increased weight of chain may be problematic
  - If wear flats reduce rolling between links, bending loads increase



Vicinay Cadenas



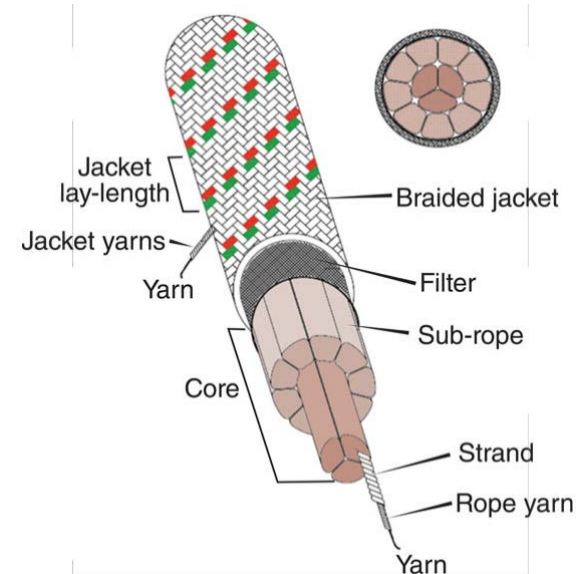
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# *Possible* Solutions to mooring failure

Use corrosion resistant or non-corroding materials

- Stainless steel commonly used in fish farms with high-flow
- Fibre ropes have very good track record in O&G industry
  - Polyester, Aramid, High-modulus Polyethylene (HMPE)
    - (near) Neutral weight in water
    - Low abrasion resistance – for use in water column only
    - Fibre rope can become damaged if sand gets into rope fibres
    - Unknown how sand filters will perform in high flow like Minas Basin, but commonly used in Gulf of Mexico where loop currents can reach 1.8-2m/s



# *Possible* Solutions to cable failure

External coatings – eliminate water getting to armour strands

- Plastic sheathing
  - May get damaged from abrasion on hard seabed
- Grease or blocking compound on armour wires, with or without additional plastic sheathing
  - Most effective in cold water (warm water lowers the viscosity)





# CONTACT

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